Lipids of physiological significance

Lipids I



Lipids

heterogeneous group of compounds, that are

- insoluble in water
- soluble in nonpolar solvents





LIPIDS



Phospholipid



Glycolipid



Steroid





Classification of lipids (according to physiological value)



FUNCTIONS OF LIPIDS

- Energetic
- Structural
- Transport
- Thermal insulator
- Electrical insulator
- Hormonal
- Protection of organs and tissues from mechanical damage
- Dietary lipids source of fat-soluble vitamins and essential fatty acids
- Source of water



Lipids of human tissues

10-12 % of body weight (2-3 kg structural lipids, other – reserve)

- Adipose tissue: 75 % of dry weight
- Nervous tissue: 50 % of dry weight (30% sphingomyelins, 10% cholesterol, 7% gangliosides and cerebrosides)
 Liver: 10-14%

Fatty acids of human tissues: classification, representatives

- Saturated (palmitic, stearic)
- Monounsaturated (palmitoleic, oleic)
- Polyunsaturated (linoleic, linolinic, arachidonic)

Saturated	None	
Monounsaturated	One	
Polyunsaturated	Multiple (>1)	

Types of Fatty Acids



 $\bigcirc = C = 0 = H$

Saturated fatty acid



Monounsaturated fatty acid



Polyunsaturated fatty acid



Functions of Fatty Acids

1) Fatty acids are building blocks of phospholipids and glycolipids.

 Many proteins are modified by the covalent attachment of fatty acids, which target them to membrane locations

 Fatty acids are fuel molecules. They are stored as triacylglycerols. Fatty acids mobilized from triacylglycerols are oxidized to meet the energy needs of a cell or organism.

4) Fatty acid derivatives serve as hormones and intracellular messengers

Omega-3 fatty acids



Omega-6 fatty acids



Essential Fatty Acid Pathways

OMEGA-6 Familiy

LINOLEIC ACID (LA) Eg. Soybeen, Cunflower, Corn, Canola + Rice Barn Oil **OMEGA-3** Family

ALPHA -LINOLENIC ACID (ALA) Eg. Flaxseed, Hemp +Chia Seed Oil

Your Body Converts LA to:

GAMMA-LINOLENIC ACID (GLA) Eg. Evening Primrose Oil + Borage Oils

ARACHIDONIC ACID (AA)

Eg. Meat

EPA Inhibits this Conversion

Your Body Converts ALA to:

EICOSAPENTANONIC ACID (EPA) Eg. Fish Oil

/ Your body converts EPA to:

DOCOSAHEXAEONIC ACID (DHA) Eg. Fish Or Algae Oil

PRO-INFLAMMATORY PRODUCTS

ANTI-INFLAMMATORY PRODUCTS

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Dietary lipids



70-140 g per day

(depend on age, sex, physical work and climate)

10 % oils

With a balanced diet fats should provide no more than 30% of total calories

Impact of a meal rich in saturated fatty acids



Digestion of lipids

The main cleavage site of lipids is the small intestine.

To increase the contact surface with hydrophilic enzymes fats should be emulsified

(break into small drops).

Emulsification takes place under the action of bile salts. Peristalsis and bubbling CO₂ also promote emulsification.



MICELLE FORMATION



Absorption of lipids



Digestion of lipids

 Phospholipids are hydrolyzed by pancreatic phospholipase A₁, A₂, C, and D.



 Esters of Cholesterol are digested by pancreatic cholesterol esterase

Absorption of lipids in the gastrointestinal tract



Re-esterification inside mucosal cell to form triacylglycerol (resynthesis of triacylglycerol).



Resynthesis of triacylglycerols Formation and release of chylomicrons

In epithelial cells, triglycerides are re-formed and, along with other fats, are enclosed by a membrane from the SER. They are coated with proteins to form chylomicrons and enter lacteals to be transported to the blood. Lacteal Fatty acids and Triglycerides monoglycerides and other fats Proteins SER Chylomicron / Small intestine epithelial cell



A shematic diagram of a chylomicron



Metabolism of Chylomicrons



Disorders in digestion and absorption of lipids in the gastrointestinal tract

In all these disorders the feces contain much unsplit fat or unabsorbed higher fatty acids

this state referred as steatorrhea

Normally less than 5 % of ingested fat is excreted in faeces.

steatorrhea

- Pancreatogenic
- (deficient supply of pancreatic lipase to the intestine)
- Hepatogenic
- (disordered delivery of bile to the intestine)
- Enterogenic (deseased GI tract)

Oxidation of fatty acids

- Occurs in mitochondria
- The fatty acids are first converted to their acyl coenzyme A derivatives





RCOOH + HSCoA + ATP RCO~CoA + AMP + PP_i

enzyme: acyl-CoA-synthetase (found in the endoplasmic reticulum, peroxisomes, mitochondria)

CARNITINE





Role of carnitine in the transport of long-chain FA



Carnitine deficiency

Deficiency of carnitine can occur:

(a) In newborns: specially premature infants, owing to inadequate synthesis or renal leakage.(b) In adults:

losses can occur in hemodialysis

in patients with organic acidurias, carnitine is lost in urine being conjugated with organic acid.

Treatment: oral therapy with carnitine.

Oxidation of fatty acids

In β-oxidation two carbons are cleaved from acyl-CoA molecules, starting at the carboxyl end, forming acetyl-CoA

A new acyl-CoA is two carbons shorter than the original acyl-CoA

The pathway produces FADH₂ and NADH

(Harper's Illustrated Biochemistry)



From Mathews and van Holde: Biochemistry 2/e. © The Benjamin/Cummings Publishing Co., Inc.



ATP Formation in the oxidation of palmitic acid

		- 1(2)
TOTAL		131 (108)
8 Acetyl-CoA	x 12 (10)	96 <mark>(80)</mark>
7 FAДH ₂	x 2 (1.5)	14 <mark>(10.5)</mark>
7 NADH	x 3 (2.5)	21 (17.5)

Oxidation of fatty acids with odd number of carbon atoms

- Fatty acids with an odd number of Carbon atoms are oxidised by β-oxidation pathway to produce acetyl-CoA until a 3carbon residue propionyl-CoA is left.
- Propionyl-CoA is metabolised to succinyl-CoA through methyl malonyl-CoA.

Oxidation of fatty acids with odd number of carbon atoms

propionyl-CoA $\rightarrow \rightarrow \rightarrow$ succinyl-CoA / citric acid cycle



β-oxidation of unsaturated fatty acids

- β -oxidation occurs normally for 3 H_sc, rounds until a cis- Δ^3 -enoyl-CoA is formed.
- Acyl-CoA dehydrogenase can not add double bond between the α and β carbons.
- Encyl-CoA isomerase converts this to trans- Δ^2 encly-CoA
- Now the β-oxidation can continue on the hydration of the trans-Δ²-enoyl-CoA
 - Odd numbered double bonds handled by isomerase



Oxidation of unsaturated fatty acids

$\Delta^3 cis \rightarrow \Delta^2 \text{-}trans$ -enoyl-CoA isomerase

Lipid peroxidation is the result of damaging effect of reactive oxygen species (ROS) on polyunsaturated fatty acids which are constituents of membrane phospholipids

Lipid peroxidation is a chain reaction providing a continuous supply of ROS that initiate further peroxidation and thus has potentially devastating effects. The whole process can be depicted as follows:

Initiation
$$RH \rightarrow H^{\bullet} + R^{\bullet}$$

Propagation $\stackrel{\bullet}{\rightarrow} R^{\bullet} + O_{2} \rightarrow ROO^{\bullet}$
 $ROO^{\bullet} + RH \rightarrow ROOH + R^{\bullet}$
Termination $\begin{cases} ROO^{\bullet} + R^{\bullet} \rightarrow ROOR \\ ROO^{\bullet} + ROO^{\bullet} \rightarrow ROOR + O_{2} \\ R^{\bullet} + R^{\bullet} \rightarrow RR \end{cases}$ Formation of stable products







The reaction is initiated by an existing free radical (X•, R•), by light, or by metal ions. Malondialdehyde is only formed by fatty acids with three or more double bonds and is used as a measure of lipid peroxidation together with ethane from the terminal two carbons of ω 3 fatty acids and pentane from the terminal five carbons of ω 6 fatty acids.

Peroxidation of lipids exposed to oxygen is responsible not only for deterioration of foods (rancidity), but also for damage to tissues *in vivo*, where it may be a cause of cancer, inflammatory diseases, atherosclerosis, and aging

The deleterious effects are considered to be caused by free radicals, molecules that have unpaired valence electrons, making them highly reactive

To control and reduce lipid peroxidation, both humans in their activities and nature invoke the use of antioxidants.

Propylgallate, butylated hydroxyanisole, and butylated hydroxytoluene are antioxidants used as food additives. Naturallyoccurring antioxidants include vitamin E, which is lipid soluble, and urate and vitamin C, which are water soluble. Beta-carotene is an antioxidant at low PO₂.

Antioxidants fall into two classes:

- (1) preventive antioxidants, which reduce the rate of chain initiation
- (2) chain-breaking antioxidants, which interfere with chain propagation.