BIOCHEMISTRY OF THE KIDNEY AND URINE



Doc. Dr. Anna Vinitskaya Department of Biological Chemistry



Kidney, biochemical functions, metabolism of the kidney. Role of kidney in regulation of pH balance

Morphology of the urinary system

- Organ system that produces, stores, and carries <u>urine</u>
- Includes two <u>kidneys</u>, two <u>ureters</u>, the <u>urinary</u> <u>bladder</u>, two sphincter muscles, and the <u>urethra</u>.
- Humans produce about <u>1.5 litres of urine</u> over 24 hours, although this amount may vary according to the circumstances.
- Increased fluid intake generally increases urine production.
- Increased <u>perspiration</u> and <u>respiration</u> may decrease the amount of fluid excreted through the kidneys.



- Is the paired organ (weight about 300 g)
- Its <u>parenchyma</u> is divided into two major structures:
 - -the outer renal cortex
 - -the inner <u>renal medulla</u>.
- Consists of about 1 million filtering units termed <u>nephrons</u> (basic structural and functional unit)

RENAL FUNCTIONS

- 1. Formation of urine and excretion the waste metabolic products from the body *(ammonium, urea, uric acid, creatinine, etc.).*
- 2. Reabsorption and preservation of vital nutrients (glucose, proteins, amino acids, etc).
- 3. Regulation of water and electrolyte balance (Na+, K+, Cl-)

RENAL FUNCTIONS (Cont.-d)

- 4. Maintenance of pH homeostasis in the blood plasma (via renal mechanisms).
- 5. Regulation of blood pressure (via maintaining the electrolyte and water balance, role of renin).

RENAL FUNCTIONS (Cont.-d)

- 6. Metabolic function: *synthesis of specific proteins and other compounds; catabolism of lipids, carbohydrates, energy metabolism.*
- 7. Endocrine function: *synthesis of hormone-like substances*
 - renin
 - Erythropoietin
 - 1-α,25-dihydroxycholecalcipherol



- is <u>the enzyme</u> that is synthesized by the juxtaglomerular cells in the kidneys from prorenin, already present in the blood.
- It is the part of the <u>RENIN–ANGIOTENSIN SYSTEM</u> (<u>RAS</u>) and that regulates <u>blood pressure</u> and <u>fluid</u> <u>balance</u>.
- <u>renin</u> is then secreted into the <u>circulation</u> and catalyzes conversion of <u>angiotensinogen</u> into <u>angiotensin-I.</u>

Erythropoietin

- Polypeptide hormon that is formed predominantly by the kidney (also by the liver)
- It controls the differentiation of the bone marrow stem cells.
- The release is stimulated by hypoxia (low pO₂).
- The hormon ensures that the bone marrow cells are converted to erythrocytes, so that their concentration in the blood increases (erythropoiesis)



- steroid-related hormon involved in calcium homeostasis.
- active form of VITAMIN D₃ (CALCITRIOL)
- increases the level of Ca²⁺ in the blood
- increases the uptake of Ca²⁺ from the intestine into the blood
- increases reabsorption of Ca²⁺ by the kidneys

2. Metabolism in kidneys

- The physiological processes that occur in the kidney require a large supply of energy.
- High intensity of energy metabolism

THE MOST ACTIVE METABOLIC PATHWAYS

- aerobic glycolysis, gluconeogenesis
- β-oxidation of fatty acids
- Utilization of ketone bodies
- Transamination and deamination of amino acids.
- Energy dependent transport of ions, nutrients, waste products, etc (filtration and reabsorption).

Fate of glucose in the kidneys



Carbohydrate metabolism in the kidney

- Renal cortex more aerobic glycolysis,
- Renal medulla more anaerobic glycolysis

Gluconeogenesis

- Take place <u>in renal cortex</u> only
- The main substrate is <u>glutamine</u>
- Other substrates are amino acids, lactate, glycerol or fructose (all are obtained from the blood plasma)
- The ammonia derived from this glutamine serves to buffer the pH of the urine.
- Gluconeogenesis is regulated by insulin, adrenalin, and induced by cortisol.

Energy metabolism

- Most metabolic processes in the kidney are aerobic, and oxygen consumption in the renal tissue is very high.
- The kidney consumes about 10% of all O₂ in the body;
- About 70% of all that O₂ is used for generation of energy for <u>maintaining of</u> <u>tubular reabsorption of glucose and amino</u> <u>acids.</u>

Energy sources of the kidney

- About 90% of ATP is derived from β -oxidation of fatty acids and utilization of ketone bodies.
- The rest 10% of ATP is derived from oxidation of glucose, lactate, pyruvate, glycerol, citrate and amino acids absorbed from the blood.
- Most of all ATP available in the renal tubular cells is used for functioning of Na⁺/K⁺-ATPase responsible for reabsorption of glucose and amino acids.

Lipid metabolism in the kidney

- From all lipids
 - 80-85% are phospholipids and cholesterol
 - 10-15% are TAGs
- Higher rate of β- oxidation of fatty acids and utilization of ketone bodies
- Place of utilization of plasma lipoproteins.

Metabolism of amino acids in the kidney

- Regulation of plasma concentrations of amino acids
- About 70 g of amino acids per day, derived from the diet and metabolism in the liver, muscle and other tissues, are filtered from the arterial blood by the kidney. Most of these amino acids are <u>actively</u> <u>reabsorbed in the proximal tubules</u> and after metabolism leave the kidney by the renal vein.
- The most metabolized amino acids are **Glutamine**, **Glutamate**, **Aspartate**, **Glycine**.
- Glutamine is used for regulation of pH of blood plasma.

Metabolism of amino acids in the kidney (cont.-ed)

- High activity of transaminases and glutamate dehydrogenase (transamination and deamination of amino acids).
- Ammonia produced in the kidneys is used for regulation of the acid-base balance.

First reaction in the synthesis of creatinine.
Arginine + glycine → guanidine acetate

Metabolism of proteins in the kidney

- The kidney is a major site for the catabolism of both circulating and kidney peptides and proteins
- Of circulating proteins the kidney <u>breaks down</u> <u>small and medium-sized blood plasma proteins</u> (below 6000 Da), such as <u>insulin and peptide</u> <u>hormones.</u>

MECHANISMS FOR REGULATION OF pH

- The acid-base balance in the blood and extracellular fluids is maintained by cooperative interaction of three main mechanisms:
 - buffer systems of the blood;
 - respiratory mechanisms,
 - renal mechanisms.
- The effect of buffer systems of the blood is manifested <u>within 30 seconds</u>.
- The lungs require a period <u>of 1-3 min</u> to normalize pH in the blood.
- Kidney needs <u>10-20 h</u> to restore a disturbed acidbase balance.

Renal mechanisms of acid-base homeostasis (pH balance)

The kidney regulates pH of the blood plasma by:

- Reabsorption of bicarbonate ions (HCO₃⁻)
- Excretion of hydrogen ions (H⁺)
- 3 mechanisms provide excretion of excessive hydrogen ions from blood:
 - I. Reabsorption of bicarbonate ions
 - II. Excretion of H⁺ via phosphate buffer system (H₂PO₄⁻/ HPO₄²⁻)
 - III. Excretion of H^+ as ammonium ion (NH_4^+)

Mechanism 1. Formation of carbonic acid with the following dissociation to H⁺ and HCO₃⁻ in the proximal convoluted tubules **Tubular** Blood **Tubular lumen** cells Na⁺ Na⁺ Na+◄ H+ ► H+ H+⁻ HCO HCO₃- H_2CO_3 H+ H+ $H_2O \perp CO_2$ H+ H+ H+

- 1. CO₂ and H₂O are produced in the mitochondria of tubular cells.
- 2. Next the enzyme <u>carbonic anhydrase</u> binds CO_2 to H_2O to form H_2CO_3
- 3. Carbonic acid easily dissociates to give HCO_3^- and H^+ .
- 4. H⁺ from the blood and tubule cells are secreted to the lumen in exchange for Na⁺.
- 5. As the result of these events hydrogen ions (H⁺⁾ are eliminated from organism, and HCO₃⁻ is reabsorbed in plasma along with Na⁺.
- 6. THIS MECHANISM SERVES TO INCREASE THE ALKALI RESERVE OF THE BLOOD PLASMA AND TO PREVENT THE LOSS OF BICARBONATE WITH THE URINE.

Mechanism 2. Hydrogen phosphate ion is converted into dihydrogen phosphate in the distal convoluted tubules



- 1. The renal filtrate produced in the glomeruli contains Na₂HPO₄.
- In acidosis excessive amounts of H⁺ are secreted from blood and the tubular cells in exchange for Na⁺.
- 3. In tubular lumen dihydrogen phosphate (HPO₄⁻²) reacts with Na⁺ and H⁺ ions to form NaH₂PO₄.
- 4. Sodium dihydrophosphate (NaH₂PO₄) is then excreted through the urine.
- 5. <u>THIS MECHANISM ALLOWS TO REABSORB AND</u> <u>SAVE SODIUM IONS AND AVOIDS DEHYDRATION.</u>

Mechanism 3. Excretion of hydrogen ions as ammonium salts in the distal tubules



- 1. In acidosis the uptake of glutamine and H⁺ from blood to the tubule cells is increased.
- 2. Also acidosis stimulate activity of the kidney <u>glutaminase</u> and <u>glutamate dehydrogenase</u>, the enzymes that produce NH₃ from glutamine and glutamate.
- 3. Non-ionized ammonia can easily diffuse through cell membrane into the tubular lumen. In turn H⁺ diffuse into the lumen in exchange for Na⁺.
- 4. In the tubular lumen NH_3 binds with H^+ to form ammonium ion (NH_4^+), the latter is neutralized by Cl^-
- 5. Resulted <u>AMMONIUM CHLORIDE</u> is excreted through urine from the body.

II.

General characteristics and composition of urine. Pathologic components of urine. Role of urine analysis in diagnosis.

Urine formation

- For the production of urine, the kidneys do not simply pick waste products out of the bloodstream and send them along for final disposal.
- The kidneys' 2 million or more nephrons (about a million in each kidney) form urine by
 - **3 precisely regulated processes**
 - **1. glomerular filtration**
 - 2. reabsorption
 - 3. secretion

PHYSICAL CHARACTERISTICS OF THE URINE USED IN CLINICAL LABORATORIES

include appearance, color, turbidity (transparency), smell (odor), pH (acidity alkalinity) and volume. Many of these characteristics are notable and identifiable by vision alone, but some require laboratory testing.

1. APPEARANCE (transparency):

Normal urine is clear and transparent

 \succ Cloudy \rightarrow presence of oxalates, protein, cell elements, bacteria, etc.

2. SPECIFIC GRAVITY (OR DENSITY) .

is the ratio of the weight of a volume of the urine compared with the weight of the same volume of distilled water.

Normal SG is most commonly <u>1.015-1.029</u>.

More concentrated urine, dehydration, excessive sweating, urinary tract infections.

↓ More diluted urine in renal failure, pyelonephritis, diabetes insipidus.

3. ODOUR:

- Smell of acetone: Diabetic ketoacidosis
- Fishy: Presence of blood

4.COLOUR:

Is Amber-yellow due to urochrome

- \succ High colour \rightarrow Jaundice
- **Red colour** \rightarrow **Blood**
- ➢ Dark colour → dehydration
- $\succ \text{ Light colour} \rightarrow \text{drinking}$
more water



5. pH of the urine:

Ranges from 5.3 to 7 with an average of 6.2

- high <u>protein</u> diets, alcohol → more acidic urine,
- vegetarian diets \rightarrow more alkaline urine. Severe shifts in the pH values:
- Metabolic acidosis, diabetes mellitus, starvation, fever
- **1** Metabolic alkalosis, cystitis, pyelitis.

6. VOLUME OF THE URINE

Normal - in woman - 1200 ml/day, in men - 1500 ml/day. **D** Polyuria: \uparrow Urine output \rightarrow > 2.5 litres/day Seen in: - increased water ingestion - diabetes mellitus and insipidus \Box Oliguria: \downarrow Urine output \rightarrow 300 to 500 ml/day Seen in: - dehydration - acute glomerulonephritis **Anuria:** $\downarrow \downarrow \downarrow \downarrow$ Urine output $\rightarrow 0 - 50 \text{ ml/day}$ Seen in: - renal shutdown in acute renal failure

Daily diuresis

is increased urination and the physiologic process that produces such an increase. It involves extra urine production in the kidneys as part of the body's homeostatic maintenance of fluid balance

Composition of the Urine

- A sterile fluid composed of:
 - 1.Water (about 95%)
 - 2.Organic constituents (35-45 g/day)
 - 3. Inorganic constituents (15-25 g/day)



Also pigments, hormones and their derivatives, etc.

Inorganic components of urine are the same as in the blood plasma	
	mmol/day
Cl-	120 – 240
Na ⁺	100 – 150
K +	60 - 80
SO ₄ ²⁻	30 - 60
NH ₄ ⁺	30 - 60
HPO ₄ ²⁻	10 - 40
Ca ²⁺	4 - 11
Mg ²⁺	3 - 6

Pathological components of the urine

- are <u>absent</u> is the normal urine or present <u>ONLY in trace amounts</u> and increase in diseases.
 - Proteins (proteinuria)
 - Glucose (Glucosuria)
 - Ketone bodies (ketonuria)
 - Occult blood (hematuria)
 - Bile pigments.
 - Porphyrins

Causes of Proteinuria

- There are three main mechanisms to cause proteinuria:
 - 1. Due to disease in the **glomerulus** (renal proteinurea)
 - 2. Because of increased quantity of proteins in <u>serum</u> (overflow proteinuria)
 - Due to low reabsorption at <u>proximal</u> <u>tubule</u> (Fanconi syndrome) (extrarenal proteinurea).
 Other causes:
- Action of some medications, used in cancer treatment,
- Excessive fluid intake (drinking in excess of 4 liters of water per day)



- Is the excretion of glucose into the urine
- Detectable glucose in urine appears <u>WHEN THE</u> <u>BLOOD GLUCOSE LEVEL EXCEEDS RENAL THRESHOLD</u> (>10 MMOL/L)

• Main causes

- Elevated blood glucose level due to different reasons (diabetes mellitus).
- <u>Renal glycosuria</u> due to an intrinsic problem with glucose reabsorption within the kidneys themselves.

Ketonuria

- Is a medical condition in which <u>ketone bodies</u> are present in the urine
- Main causes:
 - <u>Diabetes</u>, renal <u>glycosuria</u>, or <u>glycogen</u> storage disease.
 - Dietary conditions such as <u>starvation</u>, fasting, high protein, or low carbohydrate diets,
 - prolonged <u>vomiting</u>,
 - <u>anorexia</u>.
 - Conditions in which metabolism is increased, such as <u>hyperthyroidism</u>, fever, pregnancy or <u>lactation</u>.



Bile pigments in urine

- Breakdown products of normal heme catabolism, caused by the body's clearance of aged <u>red blood cells</u> that contain <u>hemoglobin</u>.
- <u>Direct bilirubin</u> is present in the urine both in obstructive and hepatocellular jaundice.
- <u>Urobilinogen</u> increases in hepatocellular jaundice.
- <u>Stercobilinogen</u> is absent in the urine in obstructive jaundice due to obstruction of the bile duct.

Clinical Urine Tests

- are various tests of urine for diagnostic purposes.
- <u>Macroscopic Examination</u> (including naked-eye (gross) examination for color and smell)
- <u>Microscopic Examination</u> (presence of RBC, WBC, etc).
- <u>Cytological Examination (microbiological</u> culture of urine)
- <u>Chemical Analysis (Urine analysis or Urinalysis</u>)

Urine Analysis or Urinalysis

- Fresh Sample = Valid Sample
- The target parameters that can be measured or quantified in urinalysis include naked-eye (gross) examination for COLOR and SMELL plus analysis for many substances and cells, as well as other properties, such as <u>SPECIFIC</u> <u>GRAVITY</u>.
- A part of a urinalysis can be performed by using <u>URINE TEST STRIPS</u>, in which the test results can be read as color changes.

Urine test results should always be interpreted using the reference range of some laboratory tested compounds and enzymes

- Amylase 28 160 g/h.L
- comes from pancreas
- in acute pancreatitis
- Uric acid 1.6 6.4 mmol per day
- End product of purine metabolism
- ♦ ↑ in leukemia
- \downarrow in gout, nephritis
- Urea 333 583 mmol per day
- End product of protein metabolism
- Ievels is seen in kidney and liver dysfunction

Creatinine – 1-2 gram per day (male) 0.5 – 1.6 g per day (female)

Anhydride form of creatine formed in muscles

Urinalysis test strips



• A urine sample is tested for several components simultaneously by dipping a test strip into a sample and comparing the reagent areas to a color chart

- A urine specimen is often collected from a patient entering a hospital and is checked with a URINE TEST STRIP that contains bands of different reagents.
- These reangents may react with particular pathological components, and patogens, and also include testing of pH or Specific Gravity.



Case 1

Diabetes





