Bioorganic Chemistry Academic plan of laboratory practical classes for students of FS faculty II semester of 2018-2019 st. year

- 1. Introduction. Spatial structure of organic molecules (configurations and conformations). Principles & rules of chemical nomenclature.
- 2. Stereoisomerism, its importance for biological activity of organic compounds.
- 3. Electronic structure of chemical bonds and influence of atoms in organic molecules.
- 4. Reactions mechanisms in Organic chemistry. Reactive ability of aliphatic hydrocarbons (S_R and A_E reactions).
- 5. Reactions mechanisms in Organic chemistry. The reactive ability of aromatic hydrocarbons (S_E reactions).
- 6. Acid-base properties of organic compounds. Reactions of oxidation.
- 7. Nucleophilic substitution (S_N) on saturated carbon atom.
- 8. Biologically important reactions of carbonyl compounds (aldehydes and ketones) (A_N reactions).
- 9. Carboxylic acids & their functional related compounds.
- 10. Saponifiable (hydrolysable) lipids. The structure of biological membranes.
- 11. Heterofunctional aliphatic compounds components of metabolic processes. Structure & reactivity of hydroxy-, oxo-, and amino acids. Stereochemistry.
- 12. Biologically active heterocyclic aromatic & heterocyclic compounds, metabolites, and bioregulators.
- 13. Carbohydrates. Monosaccharides.
- 14. Oligo- and polysaccharides.
- 15. Amino acids.
- 16. Peptides, structure, properties, biological importance. Levels of protein molecules organization. Structure and function of hemoglobin.
- 17. Purine and pyrimidine bases. Nucleosides. Nucleotides. The influence of environmental factors on the structural components of nucleic acids.
- 18. Physiologically active heterofunctional benzene-related compounds and derivatives.

Head of General and Bioorganic Chemistry Dept. Associate Professor

V.V. Boltromeyuk